

## Crystallographic report

A new polymorph for bis[bis(*N,N*-dibenzylthio-carbamato)cadmium(II)]M. Saravanan,<sup>1</sup> K. Ramalingam<sup>1\*</sup>, G. Bocelli<sup>2</sup> and R. Olla<sup>2</sup><sup>1</sup>Department of Chemistry, Annamalai University, Annamalaiagar 608 002, India<sup>2</sup>IMEM-CNR, Parco area delle scienze, Parma 43100, Italy

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The title compound is a centrosymmetric dimer with each cadmium in a distorted  $\text{CdS}_5$  square pyramidal geometry. The Cd–S bond distances range from 2.5626(11) to 2.8459(11) Å. Copyright © 2004 John Wiley & Sons, Ltd.

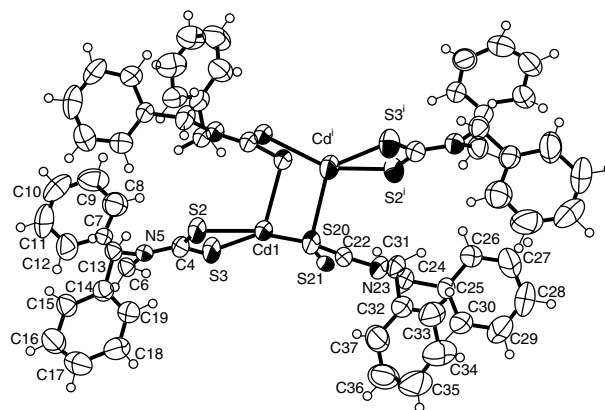
KEYWORDS: cadmium; dithiocarbamate; crystal structure

## COMMENT

The main-group elements zinc, cadmium and mercury readily form dithiocarbamate complexes.<sup>1</sup> Two independent and centrosymmetric molecules comprise the asymmetric unit in the title compound,  $[\text{Cd}(\text{dbzdtc})_2]_2$ ; one molecule is shown in Fig. 1. The cadmium atom is in a distorted square pyramidal coordination environment with Cd–S bonds in the range 2.5626(11) to 2.8459(11) Å; as a consequence, the C–S bonds are asymmetric. The mean thioureide bond distance is 1.344(4) Å. The crystal structure of a monoclinic polymorph of  $[\text{Cd}(\text{dbzdtc})_2]_2$  has been reported recently.<sup>2</sup>

## EXPERIMENTAL

Dibenzylamine (10 mmol, 2.0 ml) in ethanol (10 ml) and carbon disulfide (10 mmol, 0.6 ml) in ethanol (10 ml) under cold condition (5°C) were mixed thoroughly and stirred. The resulting yellow dithiocarbamic acid was mixed with  $\text{Cd}(\text{NO}_3)_2$  (5 mmol, 1.5 g) in water (5 ml) with constant stirring. The precipitated complex was filtered, washed with water, alcohol and was then dried in air. Anal. Found: C, 53.9; H, 4.0; N, 3.9. Calc. for the complex: C, 54.9; H, 4.2; N, 4.2%. Electronic spectrum: CT band at 280 nm. Thioureide band at  $1527\text{ cm}^{-1}$  (IR). Suitable single crystals were obtained from benzene solution. The intensity data were collected on a Bruker AXS SMART CCD diffractometer with a colourless crystal  $0.19 \times 0.26 \times 0.39\text{ mm}^3$ .  $\text{C}_{60}\text{H}_{56}\text{N}_4\text{S}_8\text{Cd}_2$ ,  $M = 1314.37$ , triclinic,  $P\bar{1}$ ,  $a = 10.226(2)$ ,  $b = 16.545(3)$ ,  $c = 17.882(2)$  Å,  $\alpha = 81.36(2)^\circ$ ,  $\beta = 80.45(2)^\circ$ ,  $\gamma = 85.66(2)^\circ$ ,  $V = 2945.7(9)\text{ Å}^3$ ,  $Z = 2$ , 8105 unique reflections ( $\theta_{\text{max}} = 28.1^\circ$ ).  $R = 0.037$ ,  $R_w = 0.085$ ,  $\rho =$



**Figure 1.** Selected bond lengths (Å) and bond angles ( $^\circ$ ) for one of the independent molecules of  $[\text{Cd}(\text{dbzdtc})_2]_2$ : Cd1–S2 2.5834(11), Cd–S20 2.6467(11), Cd1–S21 2.5626(11), Cd1–S3 2.6158(11), Cd–S20 2.8459(11), S2–C4 1.733(3), S3–C4 1.720(3), C4–N5 1.345(4), S20–C22 1.747(3), S21–C22 1.719(3), C22–N23 1.337(4); S2–Cd1–S21 150.94(3), S3–Cd1–S21 114.69(4), S2–Cd1–S3 69.68(3), S20<sup>i</sup>–Cd1–S21 99.39(3), S21–Cd1–S20 66.72(3), S20–Cd1–S20<sup>i</sup> 93.42(3). Symmetry operation, i:  $-x, -y, -z$ .

$1.48\text{ e}^- \text{ Å}^{-3}$ . Programs used: SHELXTL-NTV5.1. CCDC deposition number: 209 678.

## REFERENCES

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